

DITERPENES AND UMBELLIFERONE DERIVATIVES FROM *HAPLOPAPPUS DESERTICOLA*

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Key Word Index—*Haplopappus deserticola*, *H. anthylloides*, Compositae, diterpenes, labdane derivatives, *nor*-labdane, umbelliferone derivatives, dimeric umbelliferone derivative

Abstract—The aerial parts of *Haplopappus deserticola* afforded three *ent*-labdane and one labdane derivative as well as a *nor-ent*-labdane. Furthermore, in addition to three known umbelliferone derivatives, a dimeric coumarin was isolated. The structures were elucidated by high field ^1H NMR spectroscopy.

INTRODUCTION

The large genus *Haplopappus* (Compositae, tribe Astereae, subtribe Solidagininae) is distributed over North and South America. So far the investigated species mainly gave clerodanes [1–4] but also some labdanes [3], numerous flavanoids [6–9], some umbelliferone derivatives [10, 11], acetylenic compounds [12] and some more widespread compounds. We have now studied two Chilean species, *H. deserticola* and *H. anthylloides*. The results are discussed in this paper.

RESULTS AND DISCUSSION

The aerial parts of *H. deserticola* Phil. afforded germa-crene D, the diterpenes **1**, **1a**, **2** and **3** (isolated as their methyl esters), the *nor*-labdane **4**, the umbelliferone derivatives **5** [13], **6** [14] and **7** [15] as well as the dimeric coumarin **8**.

The ^1H NMR spectra of **1** and **1a** (Table 1) indicated that we were dealing with diterpenes which must be closely related to agathic acid [16, 17]. However, the carbomethoxy group at C-4 was equatorial as could be established by the observed NOE's. Thus irradiation of H-20 gave clear effects with H-19 (9%) and H-6 axial (5%). Similarly H-19 gave NOE's with H-20 (10%) and H-6 axial (5%). Furthermore, the optical rotation was opposite, indicating the presence of an *ent*-labdane. The ^{13}C NMR data (Experimental) also supported the structure. All signals could be assigned by 2D-techniques. The relative position of the free acid group in compound **1** followed from the chemical shifts of H-14 and the methoxy group.

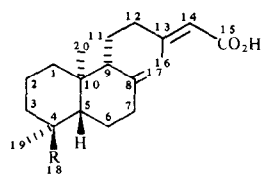
The ^1H NMR spectrum of **2a** (Table 1) showed that it was the corresponding 18-hydroxy derivative. The equatorial orientation of the hydroxy methyl group followed from the chemical shifts of H-18 and H-18'. Also the absence of a *W*-coupling supported this assumption. The enantiomer is copariferolic acid [18].

The ^1H NMR spectrum of **3a** (Table 1) was similar to those of methyl 8-hydroxy-*ent*-labd-13-*en*-15-oate [19].

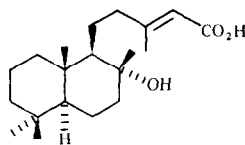
However, the optical rotation was again opposite, indicating that compound **3** was a normal labdane. The surprising co-occurrence of agathic acid and an *ent*-labdane was observed previously [17].

The ^1H NMR spectrum of **4** (Table 1) as well as the molecular formula ($\text{C}_{19}\text{H}_{30}\text{O}_3$) indicated the presence of a *nor*-labdane. Most data were similar to those of **1a** except those of the side chain. The singlet at δ 2.11 and McLafferty fragmentation [$\text{M} - \text{Me}_2\text{CO}$] indicated the nature of the degraded side chain. Accordingly, the ^1H NMR signals were close to those of the corresponding compound where the carbomethoxy group was replaced by a methyl group [20]. We have named compound **4** methyl haplodesertoate.

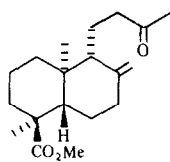
The structure of **8** could not be deduced directly from the ^1H NMR spectrum (Table 2), especially as the mass spectrum gave $\text{C}_{14}\text{H}_{14}\text{O}_3$ as highest fragment, which would be in agreement with compound **5**. As, however, the ^1H NMR data clearly indicated the absence of the coumarin double bond, a dimer of **5** was proposed. This was supported by the ^{13}C NMR spectrum (Table 2). Though only 14 signals were visible, two doublets at δ 40.3 and 36.9 required saturated carbons with only one hydrogen each. In the ^1H NMR spectrum these protons gave two triplet-like signals, which required an AA',BB'-system. It collapsed to a singlet in deuteriobenzene. Accordingly, the presence of a dimer of **5** was very likely. As photodimerization of coumarin led to four different cyclobutane derivatives [21], the stereochemistry of the dimer had to be elucidated. First the question could be solved whether the dimer was a *meso* form or a racemate. In the ^1H NMR spectrum, after addition of chiral shift reagent, the signals of the aromatic protons were split, indicating the presence of a racemic form. However, still a head-to-head *syn* dimer or a head-to-tail *anti* dimer could be present though, from the shape of the ^1H NMR signals, an *anti*-arrangement was very likely. A clear NOE between H-5 and H-4 established this assumption. As UV irradiation of coumarin gives a mixture of isomers with the head-to-head *syn* dimer as by far the main product



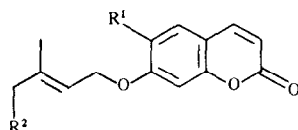
- 1** R = CO₂Me
2 R = CH₂OH



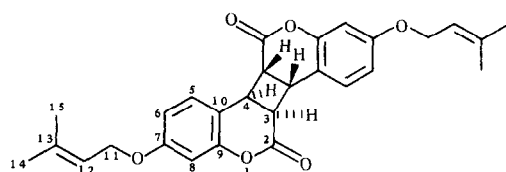
3



4



- 5** **6** **7**
R¹ H OMe OMe
R² H H prenyl



8

1a-3a are the methylesters

Table 1 ¹H NMR spectral data for **1**, **1a**, **2a**, **3a** and **4** (CDCl₃, 400 MHz, δ-values)

H	1	1a *	2a	3a †	4
5	1.66 <i>br d</i>	1.67 <i>br d</i>	1.63 <i>m</i>	0.92 <i>dd</i>	1.65 <i>br d</i>
7	2.24 <i>m</i>	2.33 <i>ddd</i>	2.36 <i>ddd</i>	1.86 <i>dt</i>	2.33 <i>ddd</i>
7'	2.02 <i>m</i>	2.01 <i>br dt</i>	1.96 <i>m</i>	1.45 <i>m</i>	2.00 <i>br dt</i>
9	1.93 <i>dd</i>	1.93 <i>dd</i>	‡	‡	1.94 <i>dd</i>
12	2.22 <i>m</i>	2.29 <i>ddd</i>	2.27 <i>br dt</i>	2.29 <i>br ddd</i>	2.58 <i>ddd</i>
12'	1.98 <i>m</i>	1.95 <i>m</i>	1.96 <i>m</i>	2.18 <i>br ddd</i>	2.32 <i>m</i>
14	5.67 <i>tq</i>	5.65 <i>tq</i>	5.63 <i>tq</i>	5.68 <i>tq</i>	—
16	2.17 <i>d</i>	2.16 <i>d</i>	2.14 <i>d</i>	2.16 <i>d</i>	2.11 <i>s</i>
17	4.85 <i>ddd</i>	4.85 <i>ddd</i>	4.83 <i>ddd</i>	1.14 <i>s</i>	4.83 <i>ddd</i>
17'	4.51 <i>ddd</i>	4.50 <i>ddd</i>	4.48 <i>ddd</i>		4.45 <i>ddd</i>
18	—	—	{ 3.40 <i>d</i> 3.08 <i>d</i>	0.86 <i>s</i>	—
19	1.13 <i>s</i>	1.13 <i>s</i>	0.73 <i>s</i>	0.78 <i>s</i>	1.14 <i>s</i>
20	0.71 <i>s</i>	0.70 <i>s</i>	0.69 <i>s</i>	0.77 <i>s</i>	0.69 <i>s</i>
OMe	—	3.69 <i>s</i>	3.68 <i>s</i>	3.68 <i>s</i>	—
	3.66 <i>s</i>	3.66 <i>s</i>	—	—	3.65 <i>s</i>

*H-6 1.76 *br d*, H-6' 1.43 *dq*, H-11 1.52 *m*, H-11' 1.21 *ddd*.

†H-6 1.25 *dq*, H-6' 1.63 *m*

‡Obscured.

J [Hz]. 5,6=12; 6,7=4, 6,7'=3; 6',7=7,7'=13, 6',7'=4.5, 9,11=12, 9,11'=2.5, 7,17=9,17=17,17'~1, 11,12=12,12'=13, 11',12=4, 12,14=14,16=1 (compound **1a** 6,6'=13)

Table 2 ^1H NMR spectral data for **8** (CDCl_3 , δ -values, 400 or 100.6 MHz respectively)

H	^1H NMR	+ Shift reagent	C_6D_6	C	^{13}C NMR	C	
3	4.15 br t	4.34 br s	3.31 s	2	164.3	9	151.3
4	4.25 br t	4.36 br s		3	40.3	10	108.6
5	7.02 d	7.22, 7.20 d	7.06 d	4	36.9	11	65.0
6	6.67 dd	6.74, 6.73 dd	6.61 dd	5	129.3	12	118.9
8	6.20 d	6.29, 6.28 d	6.29 d	6	103.0	13	138.6
11	4.39 br d	4.42 br d	{ 4.05 br dd 4.00 br dd	7	159.7	14	25.8
12	5.40 tqq	5.42 br t	5.32 tqq	8	112.2	15	18.2
14	1.78 br s	1.78 br s	1.53 br s				
15	1.70 br s	1.71 br s	1.35 br s				

$J[\text{Hz}]$ 3,4 = 3,4' = 3' 4 = 3',4' ~ 9, 5,6 = 8.5, 5,8 = 2.5, 11,12 = 7, 11,14 - 11,15 = 12,14 = 12,15 ~ 1 (in C_6D_6 11,11' = 17)

[21], compound **8** is most likely already formed in the living plant

The aerial parts of *H. anthyllodes* Meyen et Walp only gave *p*-hydroxyacetophenone. The chemistry of *H. deserticola* differs from that of most of the species studied so far, by the absence of clerodanes, while the umbelliferone derivatives are of the same type as in other species

EXPERIMENTAL

The air-dried aerial parts (200 g) of *H. deserticola* (collected in Dez 1988 in Chile, Region Metropolitana, voucher AH2, deposited in the Herbarium of the University of Chile, Santiago) was extracted with $\text{MeOH-Et}_2\text{O-petrol}$ (1:1:1). The defatted extract was first separated by CC and further by TLC and HPLC as reported previously [22]. The CC fractions were combined to four. The first fraction gave 30 mg germacrene D, the second one by HPLC ($\text{MeOH-H}_2\text{O}$, 9:1, always RP 8, flow rate 3 ml/min) 20 mg **1** (R_f 4.1 min) and a mixture (R_f 3.2 min) which gave by TLC ($\text{Et}_2\text{O-petrol}$, 1:3) 2 mg **1a** (R_f 0.70) and 2 mg **4** (R_f 0.50). The third CC fraction contained 1.5 g **1**. The last one was separated by Sephadex LH 20 ($\text{CH}_2\text{Cl}_2\text{-petrol-MeOH}$, 4:7:1) into two crude fractions (4/1 and 4/2). HPLC ($\text{MeOH-H}_2\text{O}$, 17:3) of fraction 4/1 gave 150 mg **6** (R_f 2.3 min), 1 g **5** (R_f 3.3 min) and a mixture (R_f 5.9 min) which gave by TLC ($\text{Et}_2\text{O-petrol}$, 3:1, two developments) 5 mg **7** (R_f 0.68) and 30 mg **8** (R_f 0.50). Fraction 4/2 showed no methoxy signal in the ^1H NMR spectrum. After addition of CH_2N_2 HPLC ($\text{MeOH-H}_2\text{O}$, 17:3) gave 150 mg **2a** (R_f 7.7 min) and 50 mg **3a** (R_f 9.3 min). The aerial parts (490 g) of *H. anthyllodes* (voucher Niemeyer 8809) only gave 5 mg *p*-hydroxyacetophenone and fatty acids. Known compounds were identified by comparing the 400 MHz ^1H NMR spectra with those of authentic material.

Methyl-ent-4-epi-agath-18-oate 1 Colourless oil, IR $\nu_{\text{max}}^{\text{CCl}_4}$ cm^{-1} 3500-2700, 1690, 1640 ($\text{C}=\text{CCO}_2\text{H}$), 1725 (CO_2R), MS m/z (rel int) 348, 230 [M] $^+$ (1) (calc for $\text{C}_{21}\text{H}_{32}\text{O}_4$ 348, 230), 333 [$\text{M}-\text{Me}$] $^+$ (4), 330 [$\text{M}-\text{H}_2\text{O}$] $^+$ (2), 299 [$330-\text{OMe}$] $^+$ (3), 271 [$299-\text{CO}$] $^+$ (12), 121 [C_9H_{13}] $^+$ (100)

Dimethyl-ent-4-epi-agathoate 1a Colourless oil, IR $\nu_{\text{max}}^{\text{CCl}_4}$ cm^{-1} 1710, 1645 ($\text{C}=\text{CCO}_2\text{R}$), 1725 (CO_2R), MS m/z (rel int) 362, 246 [M] $^+$ (1) (calc for $\text{C}_{22}\text{H}_{34}\text{O}_4$ 362, 246), 331 [$\text{M}-\text{OMe}$] $^+$ (8), 303 [$331-\text{CO}$] $^+$ (11), 288 [$303-\text{Me}$] $^+$ (8), 121 [C_9H_{13}] $^+$ (100), ^{13}C NMR (CDCl_3 , C-1-C-20) δ 36.9, 18.4, 37.9, 47.6, 55.9, 21.2, 37.7, 147.4, 49.7, 38.9, 26.7, 39.5, 160.8, 114.9, 167.2, 18.8, 106.9, 179.1, 16.5, 14.7, OMe 51.8, 50.7, [α] $_{\text{D}}^{24}$ -35.4 (CHCl_3 , c 7.95)

Methyl-ent-copaiferolate 2a Colourless oil, IR $\nu_{\text{max}}^{\text{CCl}_4}$ cm^{-1} 3650 (OH), 1715, 1645 ($\text{C}=\text{CCO}_2\text{R}$), MS m/z (rel int) 334, 251 [M] $^+$ (0.5) (calc for $\text{C}_{21}\text{H}_{34}\text{O}_3$ 334, 251), 319 [$\text{M}-\text{Me}$] $^+$ (1.8), 304 [$\text{M}-\text{CH}_2\text{O}$] $^+$ (4), 303 [$\text{M}-\text{OMe}$] $^+$ (5), 302 [$\text{M}-\text{MeOH}$] $^+$ (2), 274 [$302-\text{CO}$] $^+$ (1.6), 243 [$274-\text{CH}_2\text{OH}$] $^+$ (4), 55 (100)

Labd-8 α -hydroxy-13E-en-15-*oic* acid 3 Isolated as its methyl ester **3a**, colourless oil, IR $\nu_{\text{max}}^{\text{CCl}_4}$ cm^{-1} 3620 (OH), 1720, 1650 ($\text{C}=\text{CCO}_2\text{R}$), MS m/z (rel int) 336, 266 [M] $^+$ (5) (calc for $\text{C}_{21}\text{H}_{36}\text{O}_3$ 336, 266), 318 [$\text{M}-\text{H}_2\text{O}$] $^+$ (4), 304 [$\text{M}-\text{MeOH}$] $^+$ (22), 114 (100), [α] $_{\text{D}}^{24}$ +14 (CHCl_3 , c 0.13)

Methyl haplodesertoate 4 Colourless oil, IR $\nu_{\text{max}}^{\text{CCl}_4}$ cm^{-1} 1720 (CO_2R , CO), MS m/z (rel int) 306, 220 [M] $^+$ (28) (calc for $\text{C}_{19}\text{H}_{30}\text{O}_3$ 306, 220), 288 [$\text{M}-\text{H}_2\text{O}$] $^+$ (50), 273 [$288-\text{Me}$] $^+$ (14), 248 [$\text{M}-\text{Me}_2\text{CO}$, McLafferty] $^+$ (57), 246 [$\text{M}-\text{HCO}_2\text{Me}$] $^+$ (50), 121 [C_9H_{13}] $^+$ (100)

Dimeric umbelliferone-3,3-dimethylallyl ether (8) Colourless crystals, mp 100 $^\circ$, IR $\nu_{\text{max}}^{\text{CCl}_4}$ cm^{-1} 1760 (lactone), 1620 (aromate), MS m/z (rel int) 230, 94 [$\text{M}/2$] $^+$ (8) (calc for $\text{C}_{14}\text{H}_{14}\text{O}_3$ 230, 94), 162 [$230-\text{isoprene}$] $^+$ (98), 134 [$162-\text{CO}$] $^+$ (27), 69 [$\text{Mc}_2\text{C}=\text{CHCH}_2$] $^+$ (100)

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REFERENCES

- Silva, M. and Sammes, P. (1973) *Phytochemistry* **12**, 1755
- Bittner, M., Zabel, V., Smith, W. B. and Watson, W. H. (1978) *Phytochemistry* **17**, 1797
- Jakupovic, J., Baruah, R. N., Zdero, C., Eid, G., Pathak, V. P., Chau-Thi, T. V., Bohlmann, F., King, R. M. and Robinson, H. (1986) *Phytochemistry* **25**, 1873
- Bohlmann, F., Fritze, U., Robinson, H. and King, R. M. (1979) *Phytochemistry* **18**, 1749
- Kszut, S. O., Ulubelen, A., Clark, W. D., Brown, G. K. and Mabry, T. J. (1981) *Rev. Latinoam. Quim.* **12**, 12
- Bittner, M. and Watson, W. H. (1982) *Rev. Latinoam. Quim.* **13**, 24
- Ulubelen, A., Ayanoglu, E., Clark, W. D., Brown, G. K. and Mabry, T. J. (1982) *J. Nat. Prod.* **45**, 363
- Ayanoglu, E., Ulubelen, A., Clark, W. D., Brown, G. K., Kerr, R. R. and Mabry, T. J. (1981) *Phytochemistry* **20**, 1715
- Ates, N., Ulubelen, A., Clark, W. D., Brown, G. K. and Mabry, T. J. (1982) *J. Nat. Prod.* **45**, 189

- 10 Schwenker, G, Kloss, P and Engels, W (1967) *Pharmazie* **22**, 724.
- 11 Hocniscn, M. and Silva, M (1986) *Rev. Latinoam. Quim.* **17**, 19
- 12 Bohlmann, F, Burkhardt, T and Zdero, C (1973) *Naturally Occurring Acetylenes*, p 345 Academic Press, London
- 13 Prokopenko, A P (1966) *Rast Resurcy* **2**, 201
- 14 Herz, W., Bhat, S. V. and Santhaman, S. P (1970). *Phytochemistry* **9**, 891
- 15 Herz, W. and Kulantharvel, P (1985) *Phytochemistry* **24**, 1761
- 16 Ohloff, G (1958) *Helv Chim Acta* **41**, 845
- 17 Caputo, B and Mangoni, L (1974) *Phytochemistry* **13**, 467
- 18 Delle Monache, F d'Albuquerque, I L, Delle Monache, G. and Marini-Bettola, G. B (1970). *Ann. Chem.* **60**, 233
- 19 Caputo, R, Mangoni, L, Monaco, P., Pelosi, L and Previtera, C (1976) *Phytochemistry* **15**, 1401
- 20 Bohlmann, F, Jakupovic, J, Robinson, H and King, R M (1980) *Phytochemistry* **19**, 2769
- 21 Krauch, C. H., Farid, S. and Schenck, G. O (1966). *Chem. Ber.* **99**, 625
- 22 Bohlmann, F, Zdero, C, King, R M and Robinson, H (1984) *Phytochemistry* **23**, 1979

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(–)-SALZOL, AN ISOPIMARANE DITERPENE, AND A CHALCONE FROM *HYPTIS SALZMANII*

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Abstract—From the leaves of *Hyptis salzmanii* a new isopimarane diterpene, named (–)-salzol, and the new 4,2',6'-trihydroxy-4'-methoxychalcone were isolated, together with three known lignans, i.e. (+)-sesamin, (–)-cubebins, (–)-hinokinin, three known flavanones, i.e. (–)-isosakuranetin, (±)-sakuranetin, (+)-naringenin-7,4'-dimethylether, and *p*-methoxycinnamic acid. The structure of 20-benzoyloxy-6 β ,7 β ,8 β ,9 α -tetrahydroxyisopimar-15-ene, assigned to (–)-salzol, and the structure of the chalcone were determined on the basis of chemical and spectral data

INTRODUCTION

From plants of *Hyptis* genus several cytotoxic agents have been isolated, i.e. 4'-demethyldeoxydopodophyllotoxin, 14-methoxytaxodione, desoxydopodophyllotoxin [1, 2]. Continuing our studies on biologically active Brazilian plants, we have investigated the methanolic extract of the leaves of *Hyptis salzmanii* Benth. (Labiatae), that showed antimicrobial activity against *S. aureus*, *B. subtilis*, *M. smegmatis* and *C. albicans* [3]. Fractionation of this extract gave several already known compounds together with a new polyhydroxylated isopimarane diterpene named (–)-salzol (1) and a new chalcone (2)

RESULTS AND DISCUSSION

(–)-Salzol (1) was isolated as white needles, mp 196–197°, $[\alpha]_D^{20} -10.5^\circ$ (CHCl₃). The UV, ¹H (Table 1) and ¹³C NMR (Table 2) spectra of 1 indicated the presence of a benzoyl residue in the molecule. The hydrolysis of salzol under basic conditions gave in fact benzoic acid and the alcohol 1a ($[M]^+$ at *m/z* 354). The ¹³C NMR

spectrum of 1a (100 MHz, acetone-*d*₆, Table 2), showed the resonances of 20 carbon atoms, two of which belong to a vinyl group. The remaining signals were assigned to three tertiary methyl groups, five oxygenated aliphatic carbons (1 primary, 2 secondary, 2 tertiary), six methylenes, one methine, and three quaternary aliphatic carbons. ¹³C NMR and mass spectral data indicated the formula C₂₀H₃₄O₅ for 1a and suggested that the compound was a pentahydroxylated diterpene with an isopimarane-type skeleton [4]. HETCOR and long-range HETCOR experiments carried on 1 (Table 3) [5, 6], confirmed the supposed structure, and allowed us to locate four hydroxyl groups at the C-6, C-7, C-8 and C-9 positions. The primary alcoholic function was assigned to the C-20 position taking into account the altered chemical shift values of H₂-1, H₂-11 and H₂-20 protons in the ¹H NMR spectrum of 1a in comparison with those of 1, and also confirmed by NOE experiments. By treatment of 1 with pyridine and acetic anhydride only the monoacetyl derivative 1b was obtained. The difficulty in obtaining the expected diacetyl derivative was explained by the presence of an intramolecular hydrogen bond between the